

Nuclear Physics - MS

1.

nuclei	do not accept atoms	1
decreases		1
$m = 0.004 \text{ (kg)}$		1
$E = 0.004 \times 5200 \times 50\,000\,000$	allow a correct substitution of an incorrectly/not converted value of m	1
$E = 1.04 \times 10^9 \text{ (J)}$ or $E = 1\,040\,000\,000 \text{ (J)}$	allow a correct calculation using an incorrectly/not converted value of m	1
<p>any two from:</p> <ul style="list-style-type: none"> • to make sure the fusion process is possible • to develop an understanding of the process • to make adaptations to the process • to assess the efficiency of the process • to make predictions • assess safety risks • to assess environmental impact • set-up cost is lower (for small scale experiments) 		2

releases carbon dioxide which causes global warming	allow releases greenhouse gases	1
OR releases particulates which causes global dimming	allow which causes climate change	1
or which causes breathing problems		
OR releases sulfur dioxide which causes acid rain		
OR releases nitrogen oxides which causes breathing problems		
or which causes acid rain		

2.

148		1
D and E		1
line between B and 86 protons		1
same line between B and 222 mass number		1
can't predict which nucleus will decay next or can't predict when a (particular) nucleus will decay		1
one alpha decay would decrease proton number by 2		1
two beta decays would increase proton number by 2		1
so the proton / atomic number of the final nucleus is the same as the proton / atomic number of the original nucleus	this mark is dependent on scoring the first two marks	1

3.

<p>Any one from:</p> <ul style="list-style-type: none"> • (medical) x-rays • radiotherapy • nuclear weapons (testing) • named nuclear disaster eg Chernobyl / Fukushima / Three Mile Island. 	<p>allow CT scans</p> <p>allow nuclear fallout</p> <p>ignore radioactive / nuclear waste</p>	<p>1</p>
<p>uranium / plutonium</p>	<p>ignore any number given</p> <p>allow thorium</p>	<p>1</p>
<p>neutron absorbed by a uranium nucleus</p> <p>nucleus splits into two parts</p> <p>and (2 / 3) neutrons (are released)</p> <p>and gamma rays (are emitted)</p>	<p>allow an atom splits into two parts if 1st marking point doesn't score</p>	<p>1</p> <p>1</p> <p>1</p> <p>1</p>
<p>lighter nuclei join to form heavier nuclei</p> <p>some of the mass (of the nuclei) is converted to energy (of radiation)</p>	<p>allow specific examples</p>	<p>1</p> <p>1</p>

4.

${}_{82}^{206}\text{Pb}$		<p>1</p> <p>1</p>
<p>alpha radiation is highly ionising</p> <p>causing an increased risk of cancer</p> <p>or</p> <p>organ failure</p> <p>or</p> <p>radiation sickness / poisoning</p> <p>or</p> <p>mutation of genes / DNA</p> <p>or</p> <p>damage to cells / tissues / organs</p> <p>until the radioactive material is removed / excreted</p> <p>or</p> <p>activity of radioactive material reaches / approaches background radiation levels</p>	<p>allow kill cells</p> <p>allow all the alpha radiation is absorbed by the body</p> <p>ignore references to half-life</p>	<p>1</p> <p>1</p> <p>1</p>
<p>$\frac{414}{138} = 3$ (half-lives)</p> <p>$1.45 \times 10^{-4} \times 2 \times 2 \times 2$</p> <p>$= 1.16 \times 10^{-3}$ (g)</p> <p>or</p> <p>$= 0.00116$ (g)</p>	<p>an answer of 1.16×10^{-3} (g) scores 3 marks</p>	<p>1</p> <p>1</p> <p>1</p>

5.

<p>count rate = $\frac{819}{60}$</p> <p>count rate = 13.65</p> <p>corrected count rate = 13.35 (per second)</p>	<p>an answer of 13.35 (per second) scores 3 marks</p> <p>an answer of 13.95 (per second) scores 2 marks</p> <p>an answer of 801 (per second) scores 2 marks</p> <p>allow an answer of</p> <p>background = 0.30×60 = 18 (per minute)</p> <p>corrected count rate = $819 - 18$</p> <p>corrected count rate = <u>801 per minute</u></p>	<p>1</p> <p>1</p> <p>1</p>
<p>activity = 1250×180</p> <p>activity = 225 000 (Bq)</p>	<p>an answer of 225 000 (Bq) scores 2 marks</p>	<p>1</p> <p>1</p>
<p>yearly dose = 0.003×365</p> <p>which is $\ll 100$ (mSv)</p> <p>or</p> <p>(well) below the lowest dose with evidence of causing cancer / harm</p>	<p>allow yearly dose = 1.095 (mSv)</p>	<p>1</p> <p>1</p>

6.

<p>06.1</p>	<p>smoke absorbs / stops alpha radiation</p>	<p>allow alpha particles for alpha radiation</p> <p>alpha radiation does not reach the detector is insufficient</p>	<p>1</p>
<p>06.2</p>	<p>alpha radiation is not very penetrating or alpha radiation does not penetrate skin</p>	<p>allow alpha particles for alpha radiation</p> <p>allow alpha radiation does not travel very far (in air)</p>	<p>1</p>
<p>06.3</p>	<p>beta and gamma radiation will penetrate smoke</p> <p>no change (in the count rate) would be detected</p>	<p>allow beta and gamma radiation will not be stopped by smoke</p> <p>allow the change detected (in the count rate) would be too small</p>	<p>1</p> <p>1</p>
<p>06.4</p>	<p>(a long half-life means) the count rate is (approximately) constant or a short half-life means the count rate decreases quickly</p> <p>until 1.3 half-lives the count rate is above 80 per second or until 1.3 half-lives the count rate is above the threshold for the smoke alarm to be activated or after 1.3 half-lives the smoke alarm will be activated all the time</p>	<p>allow activity of source is (approximately) constant</p> <p>allow after 1.3 half-lives the count rate is below 80 per second</p> <p>so don't have to replace source or smoke detector is insufficient</p>	<p>1</p> <p>1</p>

06.5	Level 2: Relevant points (reasons / causes) are identified, given in detail and logically linked to form a clear account.	3–4
	Level 1: Relevant points (reasons / causes) are identified, and there are attempts at logically linking. The resulting account is not fully clear.	1–2
	No relevant content	0
	<p>Indicative content</p> <ul style="list-style-type: none"> • short half-life or half-life of a few hours • (short half-life means) less damage to cells / tissues / organs / body • low ionising power • (low ionising power means) less damage to cells / tissues / organs / body • highly penetrating • (highly penetrating means) it can be detected outside the body • emits gamma radiation 	

7.

cannot predict <u>which</u> dice / atom will 'decay'	accept answers given in terms of 'roll a 6'	1
cannot predict <u>when</u> a dice / atom will 'decay'		1
3.6 to 3.7 (rolls)	allow 1 mark for attempt to read graph when number of dice = 50	2
90		1
uranium		1
beta		1
proton number has gone up (as neutron decays to proton and e^-)		1
prevents contamination or prevents transfer of radioactive material to teacher's hands which would cause damage / irradiation over a longer time period.		1
		1